

1276. *The Crystal Structure of Trimethyl-(8-quinolinolato)-platinum(IV)*

By J. E. LYDON and MARY R. TRUTER

A three-dimensional *X*-ray crystal structure analysis has been carried out on the complex, $\text{Me}_3(\text{C}_9\text{H}_8\text{NO})\text{Pt}$. All measurements were made at $120 \pm 5^\circ\text{K}$. In an orthorhombic unit cell, $a = 15.23$, $b = 16.09$, $c = 9.48 \text{ \AA}$ all $\pm 0.02 \text{ \AA}$, space group $P2_12_12_1$, there are eight formula units, now shown to be four dimeric molecules. The platinum atoms were located from Patterson projections, and the other atoms, excluding hydrogen, by Fourier difference syntheses. After refinement by least-squares, allowing for anisotropic vibration of the platinum atoms and individual isotropic vibration parameters for the other atoms, R is 0.10.

The dimer is formed by a symmetrical pair of oxygen bridges between the Pt atoms, with $\text{Pt-O} = 2.24 \pm 0.03 \text{ \AA}$ and a separation of $3.384 \pm 0.003 \text{ \AA}$ between the platinum atoms. Each platinum atom is six-coordinated, by the two oxygen atoms, one nitrogen ($\text{Pt-N} = 2.13 \pm 0.05 \text{ \AA}$), and three methyl groups in the usual *cis* configuration ($\text{Pt-C} = 2.06 \pm 0.05 \text{ \AA}$). The halves of the molecule are related (within experimental error) by a non-crystallographic two-fold axis.

STRUCTURAL studies¹⁻³ on complexes of trimethylplatinum(IV) have, so far, all shown that the platinum atom is six-coordinated with the methyl groups in the *cis* configuration. Trimethyl-(8-quinolinolato)platinum(IV) has been prepared⁴ and shown to have a molecular weight which varies with solvent and temperature. An *X*-ray crystal structure analysis was undertaken to establish its stereochemistry. As reported briefly,⁵ it is dimeric in the solid state, maintaining six-co-ordination round the platinum atoms.

METHODS AND RESULTS

Our preliminary findings were that the space group was non-centrosymmetric with two platinum atoms in the asymmetric unit. Approximate calculations by Cruickshank's⁶ method

¹ A. G. Swallow and M. R. Truter; A. C. Hazell and M. R. Truter, *Proc. Roy. Soc.*, 1960, *A*, **254**, 205, 218.

² A. G. Swallow and M. R. Truter, *Proc. Roy. Soc.*, 1962, *A*, **266**, 527.

³ A. Robson and M. R. Truter, *J.*, 1965, 630.

⁴ K. Kite and M. R. Truter, *J.*, 1966, to be published.

⁵ J. E. Lydon, M. R. Truter, and R. C. Watling, *Proc. Chem. Soc.*, 1964, 193.

⁶ D. W. J. Cruickshank, *Acta Cryst.*, 1960, **13**, 774.

indicated that, even with measurements made at 120°K, we could not hope for a precision greater than about $\pm 0.05 \text{ \AA}$ in the positions of the carbon atoms; clearly, to ensure unequivocal location of the light atoms (other than hydrogen) a low-temperature technique was required.

Crystal Data.— $\text{C}_{12}\text{H}_{15}\text{NOPt}$, $M = 384.4$. Orthorhombic, $a = 15.23$, $b = 16.09$, $c = 9.48 \text{ \AA}$ all $\pm 0.02 \text{ \AA}$ at $120 \pm 5^\circ\text{K}$, $U = 2323 \text{ \AA}^3$, $D_c = 2.20$, $D_m = 2.17$ (by flotation), $Z = 8$, $F(000) = 1440$. Space group $P2_12_12_1$ (D_2^3 , No. 19) uniquely determined. $\text{Cu } K_\alpha$ radiation, single-crystal oscillation and Weissenberg photographs; absorption coefficient $\mu = 226 \text{ cm}^{-1}$; absorption corrections applied. No molecular symmetry required; two formula units, shown by structure analysis to be one dimeric molecule, in the asymmetric unit.

Structure Determination.—As there are two formula units in the asymmetric unit, it was necessary to locate two platinum atoms, designated Pt(1) and Pt(2) (Figure 1), and associated with each platinum atom three methyl groups, one oxygen atom, one nitrogen atom, and the nine carbon atoms of the 8-quinolinolato-ligand.

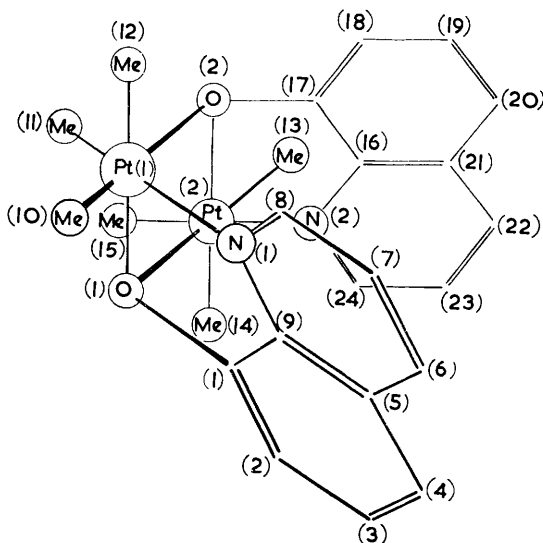


FIGURE 1. One dimeric molecule, showing the numbering of the atoms

The platinum atoms were located from Patterson projections down the three principal axes. For three-dimensional work the co-ordinates were chosen to correspond to the origin in "International Tables."⁷ Structure factors were calculated and gave an R value of 0.16, showing that the platinum atoms were correctly located. The shortest Pt-Pt distance was 3.4 \AA , indicating that the asymmetric unit consisted of a dimeric molecule probably with two bridges between the Pt atoms. A three-dimensional Fourier difference synthesis was calculated. In the resulting electron-density map, 12 peaks were readily assigned to one complete 8-quinolinolato-ligand [O(1), N(1), and C(1)—C(9) inclusive] and to O(2), with O(1) and O(2) occupying the bridging positions. Addition of these twelve atoms reduced R to 0.148. Although no more atoms were clearly shown in the electron-density map, the positions of another seven could be deduced on the assumption that the arrangement about each platinum atom was octahedral. Addition of these seven atoms, all treated as carbon atoms, reduced R to 0.119; a further difference synthesis revealed the remaining nine carbon atoms, C(16)—C(24), and indicated which of the seven was N(2). Both 8-hydroxyquinoline molecules are on the same side of the bridges joining the Pt atoms. One molecule is shown in Figure 1 and the structure as a whole in projection in Figure 2.

Refinement was carried out allowing for anisotropic thermal vibration of the platinum atoms and individual isotropic thermal parameters of the other atoms. (No attempt was made to locate hydrogen atoms.) After four cycles of refinement, R fell to 0.10, and the most significant shift, 0.06 \AA , in x for C(14), was 0.8 of the corresponding standard deviation. Each cycle

⁷ "International Tables for X-ray Crystallography, vol. I," Kynoch Press, Birmingham, 1952.

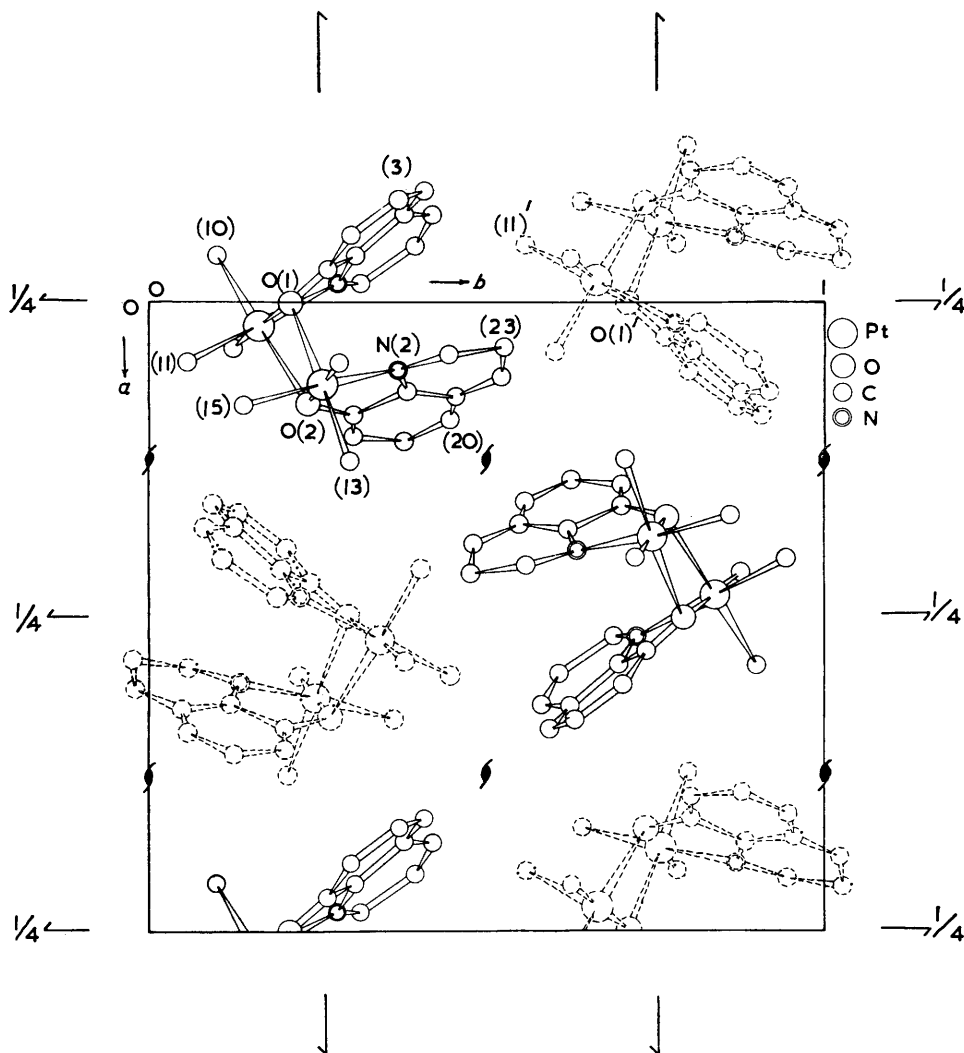


FIGURE 2. The structure projected down [001]

took 13 hours on the computer, and, as further refinement was unlikely to lead to chemically significant improvements in the accuracy of the bond lengths, it was left at this stage.

Results.—Table 1 shows the final atomic co-ordinates with their standard deviations, σ , and the isotropic vibration parameters, \bar{U} , with their standard deviations. For the platinum atoms an average value, $\bar{U} = [(U_{11}^2 + U_{22}^2 + U_{33}^2)/3]^{1/2}$ is shown for comparison with those of the light atoms. The values of the anisotropic vibration parameters in $\text{\AA}^2 \times 10^4$ are:

	U_{11}	σ	U_{22}	σ	U_{33}	σ	U_{12}	σ	U_{23}	σ	U_{13}	σ
Pt(1)	240	6	149	6	191	7	-20	6	19	8	-1	8
Pt(2)	261	7	150	6	199	7	-21	7	34	8	-34	8

In Table 2 the observed and calculated structure amplitudes are shown, and also the real and imaginary parts of the calculated structure factors.

Table 3 gives a list of the bond lengths and their standard deviations, and Table 4 lists the bond angles and their standard deviations. The molecule found has a two-fold axis of symmetry along the line through the mid-point between the platinum atoms (1.274, 3.377, 1.279) with

TABLE 1

Atomic co-ordinates and their standard deviations, in Å, and isotropic vibration parameters, U_{iso} , and their standard deviation, in Å²

	x	$\sigma(x)$	y	$\sigma(y)$	z	$\sigma(z)$	U_{iso}	$\sigma(U)$
Pt(1)	0.564	0.002	2.616	0.002	-0.058	0.002	0.020	—
Pt(2)	1.984	0.002	4.139	0.002	2.616	0.002	0.020	—
O(1)	0.00	0.03	3.38	0.03	1.95	0.03	0.020	0.006
O(2)	2.46	0.04	3.79	0.04	0.47	0.04	0.050	0.010
N(1)	-0.47	0.03	4.50	0.03	-0.48	0.03	0.016	0.006
N(2)	1.62	0.03	5.95	0.04	1.67	0.04	0.026	0.008
C(1)	-0.83	0.05	4.27	0.05	1.90	0.05	0.039	0.011
C(2)	-1.64	0.05	4.83	0.05	3.04	0.06	0.041	0.012
C(3)	-2.51	0.05	5.98	0.05	2.85	0.05	0.043	0.012
C(4)	-2.72	0.05	6.58	0.05	1.64	0.05	0.035	0.011
C(5)	-1.97	0.06	6.09	0.06	0.46	0.06	0.048	0.012
C(6)	-2.13	0.05	6.77	0.05	-0.79	0.05	0.033	0.010
C(7)	-1.38	0.05	6.36	0.05	-1.70	0.05	0.043	0.011
C(8)	-0.48	0.05	5.06	0.05	-1.60	0.05	0.044	0.011
C(9)	-1.14	0.03	4.93	0.04	0.59	0.04	0.014	0.007
C(10)	-1.17	0.03	1.63	0.04	-0.42	0.04	0.017	0.007
C(11)	1.42	0.04	0.91	0.04	0.53	0.04	0.023	0.008
C(12)	1.11	0.05	2.00	0.05	-1.96	0.06	0.045	0.012
C(13)	3.85	0.06	4.79	0.06	3.02	0.06	0.050	0.014
C(14)	1.25	0.09	4.54	0.09	4.61	0.09	0.102	0.025
C(15)	2.48	0.04	2.25	0.04	3.44	0.05	0.022	0.008
C(16)	2.13	0.04	6.17	0.04	0.37	0.04	0.019	0.007
C(17)	2.72	0.04	4.86	0.04	-0.15	0.04	0.025	0.008
C(18)	3.22	0.03	4.86	0.04	-1.41	0.04	0.014	0.007
C(19)	3.34	0.04	6.08	0.04	-2.14	0.05	0.029	0.009
C(20)	2.82	0.05	7.19	0.05	-1.70	0.05	0.033	0.010
C(21)	2.25	0.04	7.33	0.04	-0.39	0.04	0.023	0.008
C(22)	1.78	0.06	8.42	0.06	0.22	0.07	0.060	0.015
C(23)	1.07	0.04	8.49	0.04	1.44	0.04	0.020	0.008
C(24)	1.27	0.05	7.16	0.05	2.21	0.05	0.036	0.010

direction cosines -0.3 , 0.9 , -0.3 . Tables 3 and 4 are arranged to display the agreement between the corresponding dimensions in the halves of the molecule. The platinum-platinum distance in the dimer is 3.389 ± 0.003 Å.

Planes were calculated through the atoms of the 8-quinolinolato-ligands, with the results shown in Table 5. The platinum-oxygen bridge system is non-planar; the angle between the normals to the two planes III and IV in Table 5 is 12.6° .

All the intermolecular distances of less than 3.5 Å were calculated; those found were 3.45 Å between C(3) and the methyl group C(11)', and 3.40 Å between C(23) and O(1)' (see Figure 2).

DISCUSSION

The structure analysis has solved the stereochemical problem. In the solid at 120°K , and probably also at room temperature, the complex is dimeric. The structure as found satisfies the criteria for correctness¹ in that co-ordinates obtained by a mathematical process lead to reasonable bond lengths and no improbably short van der Waals separations. The methyl carbon atom, C(14), is the least satisfactorily located both in its distance from the platinum atom and in its large temperature factor, and this is reflected in the large standard deviations.

The platinum atom displays its usual six-co-ordination, achieved here by three methyl groups in the *cis* configuration, one nitrogen atom, and two oxygen atoms in a double bridge. Some caution is required in the discussion of the bond lengths because the structure, having heavy atoms in a non-centrosymmetrical space group, is one in which parameter interaction is likely to be of importance. Our method of refinement (block diagonal) does not allow for the effect of changes in the parameters of one atom on those of the other atoms, so that the standard deviations quoted so far are minimum values. Some assessment of the probable error can be made by comparing the lengths of the chemically

TABLE 2

Observed and calculated structure amplitudes

<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>			
0	0	6	139	133	0	13	1	58	64	11	48	41	2	95	95	12	47	38				
		8	119	112			3	141	139	12	77	78	3	114	114	2	4	0	105	102		
		10	96	85			6	50	40	1	5	0	4	50	40			1	57	58		
		12	50	29			7	163	154	2	1	173	5	121	141			2	136	134		
0	1	1	255	363	0	14	0	137	134	2	75	65	6	79	88			3	90	78		
		3	251	322			1	75	82	3	151	139	7	55	53			4	118	113		
		4	46	31			2	59	75	4	88	85	1	108	90			5	195	204		
		5	76	63			3	71	56	5	178	170	2	175	188			6	83	84		
		7	227	239			4	95	106	6	61	50	3	63	55			7	108	113		
		9	51	46			5	61	54	7	64	56	4	102	92			8	89	92		
		11	140	151	0	15	1	61	42	8	100	104	5	78	96			9	142	162		
		1	48	42			4	43	47	9	73	71	6	132	140			10	52	49		
0	2	2	140	133			5	74	89	10	24	27	8	93	96			11	72	61		
		3	15	15			6	35	31	11	36	37	9	66	52			12	58	48		
		4	270	288			8	66	78	1	6	0	1	149	145	2	5	0	137	146		
		5	79	66	0	16	1	77	82	1	160	160	2	70	71			1	190	212		
		7	53	54			2	151	186	2	44	32	3	115	113			2	261	276		
		8	82	83			3	64	56	3	143	125	4	74	70			3	171	172		
		9	43	28			5	52	49	4	147	140	5	108	131			4	103	102		
		10	57	61			6	107	121	5	119	114	6	54	59			5	149	143		
		11	48	42	0	17	1	141	128	6	96	79	7	50	51			6	164	172		
		1	121	114			2	41	28	7	45	44	8	61	64			7	132	142		
0	3	2	23	17			3	74	72	8	99	101	1	16	1			8	38	40		
		3	122	106			4	36	39	9	62	47	2	128	129			9	76	77		
		4	113	111			5	125	145	10	59	59	3	42	32			10	52	41		
		5	189	178			6	36	48	11	57	56	4	45	41			11	76	80		
		6	22	14	0	18	1	88	78	1	0	90	5	68	91	2	6	0	182	211		
		7	154	142			2	146	160	1	186	199	6	75	78			1	139	156		
		8	72	61			3	57	71	2	39	29	1	17	0			2	187	195		
		9	99	115			4	52	45	3	181	172	1	71	60			3	37	20		
		11	86	78	0	19	1	77	57	4	112	99	2	60	60			4	190	172		
		12	48	43	0	20	0	77	62	5	118	115	3	47	48			5	204	212		
0	4	0	79	79			1	64	68	6	75	68	4	63	68			6	147	147		
		1	64	55			2	45	49	7	112	107	5	47	48			7	64	59		
		2	270	319	1	0	1	223	261	8	129	128	6	36	26			8	91	110		
		4	68	59			2	96	82	9	60	50	1	180	104			9	144	155		
		5	74	52			3	175	189	10	43	43	1	59	57			10	98	86		
		6	191	194			4	122	132	11	71	67	3	37	39			11	61	38		
		7	49	33			5	78	62	1	0	252	4	80	80	2	7	0	112	123		
		8	29	36			6	114	105	1	134	139	5	42	46			1	138	145		
		10	60	52			7	80	76	2	147	160	1	69	61			2	229	256		
		11	47	29			8	89	86	3	118	120	1	107	91			3	162	160		
		12	68	69			10	104	96	4	205	196	3	95	102			4	75	73		
0	5	1	193	228			12	48	50	5	77	63	4	53	57			5	115	104		
		2	100	81	1	1	1	165	192	6	161	147	1	114	139			6	142	136		
		3	110	105			2	101	99	7	84	72	1	39	37			7	125	105		
		4	69	59			3	190	196	8	139	150	2	43	43			9	73	69		
		5	282	305			4	76	69	10	107	104	1	113	121			11	67	60		
		6	69	54			5	142	143	11	43	38	2	217	247			12	142	136		
		9	194	221			6	132	121	1	0	92	3	185	209	2	0	1	78	66		
		10	24	15			7	89	82	1	204	221	4	196	203			2	67	49		
0	6	0	57	53			8	57	42	2	53	40	6	153	155			3	59	54		
		3	118	105			9	78	74	3	210	226	7	204	213			4	88	74		
		6	264	261			10	108	109	4	102	95	8	107	96			5	138	134		
		10	81	90			11	44	41	5	128	115	9	81	70			6	79	61		
0	7	1	168	165	1	2	0	110	122	6	85	83	10	91	91			7	79	92		
		2	126	120			1	186	210	7	154	159	11	122	124			8	54	45		
		3	77	74			2	248	259	8	67	73	12	64	58			9	96	115		
		4	37	14			3	184	182	9	72	56	1	199	236			10	60	56		
		5	254	229			4	205	203	10	80	77	2	112	116			11	82	77		
		6	109	93			5	101	90	11	96	88	3	122	131			12	142	136		
		9	179	193			6	160	176	1	0	172	4	245	277			1	84	76		
		0	147	134			7	66	57	1	111	116	5	169	185			2	144	145		
		1	87	71			8	156	161	2	104	94	6	49	39			3	66	63		
		2	255	247			9	43	39	3	110	108	7	91	78			4	138	131		
		3	67	67			10	124	124	4	166	165	8	104	110			5	56	40		
		4	99	90			12	82	79	5	52	45	9	87	90			6	74	71		
		6	144	126	1	3	0	68	68	6	110	105	10	49	48			9	39	35		
		2	41	32			2	83	81	7	82	76	11	66	60			10	100	93		
		3	125	129			3	264	266	8	103	94	2	0	106	126			1	70	59	
		5	73	60			4	87	74	10	94	88	1	43	33			2	116	117		
		6	98	77			5	247	249	2	126	108	3	136	108			3	138	129		
		10	79	72			6	101	93	1	133	126	3	154	147			4	100	89		
0	10	0	249	258			7	139	135	2	64	65	4	107	97			5	58	50		
		1	82	76			8	87	85	3	107	109	5	130	125			6	102	107		
		3	87	79			9	133	140	4	69	54	6	76	74			7	109	119		
		4	238	226			10	71	68	5	72	65	7	166	178			8	78	84		
0	11	1	135	128			11	82	79	6	87	89	8	48	34			9	77	77		
		2	46	26			12	58	59	7	84	79	9	99	103			10	80	70		
		3	228	213	1	4	0	109	127	8	46	40	10	55	51			2	11	0	226	245
		5	64	37			1	188	217	10	83	80	11	109	121			1	164	161		
		6	80	83			2	180	170	1	81	74	12	40	37			2	63	62		
		7	236	208			3	166	149	2	95	102	2	189	185			3	157	155		
		0	209	213			4	195	187	3	99	92	3	70	63			4	161	170		
		1	93	87			5	139	132	4	65	58	4	172	159			5	122	122		
		2	58	34			6	169	164	5	53	56	5	52	33			7	87	109		
		3	100	91			7	28	19	6	80	85	6	121	102			8	62	61		
		4	208	215			8	131	146	7	64	68	7	46	39							

TABLE 2 (Continued)

<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>			
2		142	139		6		172	163		3	12	0	73	74	2		59	39	6	66	71	
3		135	140		7		150	145		1		122	121	3	136	120	4	16	1	111	99	
4		128	122		8		124	121		2		63	56	4	47	43	5		3	88	86	
5		64	47		9		94	86		3		82	87	5	151	140	4		4	109	115	
6		119	123		10		106	99		4		85	87	6	46	26	5		5	55	69	
8		103	106		11		72	69		5		90	95	7	156	158	4	17	0	119	95	
0	13	0	109	122	12		73	70		6		76	71	9	104	98	1		1	61	52	
1		115	112		3	4	0	143	144	3	13	9	66	54	10	44	37	2		61	76	
2		75	70		1		181	188		1		99	83	11	92	96	3		3	88	93	
3		103	98		2		124	110		2		109	126	4	0	214	250	4		93	93	
4		134	126		3		200	195		3		115	112	1	130	158	6		6	63	77	
5		106	95		4		198	193		4		51	46	2	133	110	0		0	157	122	
6		50	40		5		164	150		6		81	81	3	119	112	1		1	99	96	
7		66	63		6		139	126		7		106	104	4	234	237	3		3	83	93	
8		66	55		7		110	101		8		80	82	5	81	73	4		4	86	96	
9		55	53		8		134	138		0		97	129	6	60	48	5		5	56	68	
2	14	2	57	38	9		81	73		1		121	125	7	85	81	4	19	0	62	58	
3		65	64		10		98	87		2		51	44	8	98	102	2		2	37	36	
4		80	79		11		21	23		3		96	89	9	28	42	3		3	54	57	
5		65	68		10	5	0	76	56	4	17	117	119	10	40	49	4		4	61	60	
6		60	53		1		139	141		5		124	124	11	50	53	4	20	0	51	53	
7		74	84		2		120	110		6		78	65	0	186	222	1		1	53	61	
8		64	65		3		130	125		8		89	89	1	93	96	5	0	1	162	181	
0	15	0	101	101	4		88	77		3	15	1	110	98	2	136	126	2		44	41	
1		105	97		5		134	131		2		104	109	3	219	211	3		3	72	74	
2		94	111		6		148	150		3		116	128	4	165	161	4		4	119	114	
3		73	70		7		86	75		4		52	47	5	55	45	5		5	97	93	
4		67	62		8		70	64		5		63	52	6	120	122	6		6	72	54	
5		52	55		9		32	43		6		87	94	7	166	194	8		8	118	110	
6		55	68		10		100	98		7		102	100	8	74	81	0		0	101	104	
0	16	0	96	105	11		50	52		8		56	57	9	63	56	5	1	1	186	201	
1		70	46		1		152	151		0		94	104	10	68	78	2		2	134	132	
2		80	85		2		86	88		1		88	82	4	6	58	3		3	132	134	
4		94	97		3		144	140		3		72	69	3	196	179	4		4	128	160	
5		59	77		4		126	118		4		83	91	4	243	242	5		5	129	137	
6		74	83		5		92	79		5		67	82	5	134	125	6		6	104	107	
0	17	0	49	30	6		60	47		6		40	30	4	7	0	7		7	58	49	
1		131	127		7		77	68		3	17	0	48	38	1	81	94	8		110	113	
2		117	127		8		128	97		1		67	48	2	137	109	9		9	49	49	
3		116	119		0		96	85		2		56	56	3	163	159	3		3	72	64	
5		62	87		1		168	192		3		56	57	4	156	145	10		10	72	64	
6		61	70		2		45	52		6		48	56	6	74	56	5	2	0	105	92	
0	18	0	105	96	3		126	118		3	18	1	74	58	1	80	92	1		223	252	
1		72	53		4		153	151		2		57	82	2	138	134	2		2	60	61	
2		80	80		5		140	130		3		77	80	3	78	71	3		3	169	174	
4		78	88		6		98	92		3	19	0	87	97	4	171	152	4		123	115	
5		45	68		7		49	37		1		80	79	5	85	68	5		5	168	170	
0	19	0	62	36	8		107	112		3		39	42	6	50	39	6		6	92	90	
1		82	85		9		77	72		4		62	70	4	9	0	7		7	127	130	
2		69	76		10		105	99		3	20	0	44	20	1	71	49	8		113	127	
3		62	70		0		86	73		1		50	59	2	69	59	9		9	73	74	
4		36	30		1		187	205		2		75	106	3	54	40	10		10	75	79	
1	0	196	222		2		201	211		4	0	0	196	228	4	80	74	11		67	71	
2		55	45		3		165	160		1		117	120	6	51	39	6		6	214	224	
3		99	94		4		138	125		2		238	277	0	106	106	7		7	195	197	
4		28	26		5		123	112		3		117	121	1	156	155	2		2	203	184	
5		155	141		6		131	130		4		58	55	2	191	221	3		3	185	176	
6		100	99		7		129	118		5		143	134	3	116	104	4		4	215	209	
8		83	86		8		100	104		6		169	187	4	52	46	5		5	130	121	
9		48	53		10		77	69		7		123	115	6	112	117	6		6	170	162	
10		94	99		0		142	151		8		48	55	4	110	138	7		7	83	60	
3	1	0	96	94	1		177	182		9		133	116	1	127	108	8		8	126	139	
1		227	250		2		115	109		10		74	64	2	153	154	10		10	110	104	
2		95	94		3		163	151		11		71	60	3	52	39	11		11	52	45	
3		149	157		4		160	153		12		72	72	4	119	103	5		5	154	139	
4		178	186		5		147	150		4	1	0	44	37	5	134	153	1		217	236	
5		110	111		6		109	96		1		165	179	6	116	116	2		2	78	74	
6		96	92		7		76	68		2		167	159	0	57	36	3		3	188	182	
7		91	87		8		102	109		3		46	39	1	156	145	4		4	126	111	
8		139	150		9		91	96		4		113	122	2	180	214	5		5	140	126	
9		50	35		10		78	77		5		209	247	3	121	132	6		6	144	138	
10		90	75		0		59	33		6		113	120	4	47	27	7		7	131	114	
11		45	35		1		148	143		7		76	65	5	105	110	8		8	81	80	
12		109	81		2		170	175		8		93	89	6	124	132	9		9	52	62	
1	2	1	244	276	3		136	132		9		183	187	4	13	0	10		10	102	106	
2		153	152		4		88	68		10		98	86	1	80	77	11		11	68	74	
3		180	159		5		87	89		11		54	48	2	109	99	5		5	0	151	149
4		180	189		6		111	118		12		85	56	3	47	35	1		1	141	155	
5		174	175		7		88	90		4	2	0	93	91	4	53	54	2		93	86	
6		162	167		8		69	64		1		79	70	5	95	115	3		3	161	166	
7		77	73		10		71	75		2		271	273	6	88	98	4		4	112	115	
8		105	105		0		83	81		3		93	80	4	14	0	5		5	71	56	
9		81	78		1		106	104		4		114	103	1	50	52	6		6	91	88	
10		114	108		2		47	37		5		75	70	2	89	105	7		7	86	84	
11		47	44		3		88	82		6		145	145	3	65	67	8	</				

TABLE 2 (Continued)

<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>				
5	7	6	146	135	2	27	19	3	99	108	2	170	173	6	66	60	6	66	60				
		1	155	186	3	35	25	6	54	54	3	84	76	7	101	113	7	101	113				
		2	148	161	4	79	72	0	129	156	4	141	134	9	88	93	9	88	93				
		3	142	143	5	183	191	1	75	73	5	53	61	11	74	77	11	74	77				
		4	101	82	6	58	45	3	54	60	6	147	143	8	4	0	194	201	8	4	0		
		5	109	106	7	50	48	4	119	135	7	10	0	1	56	49	1	127	108	10	0	0	
		6	134	127	8	35	34	1	78	63	1	157	140	2	145	141	2	145	141	2	145	141	
		1	177	195	9	154	159	2	46	35	2	48	45	3	81	82	3	81	82	3	81	82	
		2	132	118	10	51	43	3	85	96	3	114	115	4	155	149	4	155	149	4	155	149	
		3	155	146	11	50	48	4	43	49	4	66	61	5	103	108	5	103	108	5	103	108	
		4	74	63	0	238	252	0	90	93	5	129	129	6	58	48	6	58	48	6	58	48	
		5	166	158	1	48	45	1	69	67	7	11	70	7	36	39	7	36	39	7	36	39	
		6	93	86	2	194	176	7	0	1	192	212	2	91	85	8	59	56	8	59	56		
		0	116	112	3	59	47	3	140	159	3	58	50	9	64	67	9	64	67	9	64	67	
		1	162	171	4	157	166	4	60	57	4	76	68	10	29	34	10	29	34	10	29	34	
		2	198	204	6	94	88	5	113	95	5	73	77	0	108	110	0	108	110	0	108	110	
		3	145	123	7	50	43	6	46	32	6	103	85	1	52	40	1	52	40	1	52	40	
		4	121	102	11	40	33	8	100	109	0	40	52	2	201	203	2	201	203	2	201	203	
		5	115	101	0	37	28	0	139	136	1	113	84	3	115	122	3	115	122	3	115	122	
		6	153	145	1	125	116	1	127	137	2	48	44	4	103	105	4	103	105	4	103	105	
		1	148	141	2	65	61	2	96	93	3	105	99	5	70	73	5	70	73	5	70	73	
		2	104	95	3	206	212	3	102	110	4	79	65	6	136	131	6	136	131	6	136	131	
		3	131	117	4	41	31	4	124	123	5	70	64	7	111	133	7	111	133	7	111	133	
		4	92	80	6	7	63	6	5	73	70	6	88	26	8	65	86	8	65	86	8	65	86
		5	129	131	7	159	175	6	118	117	7	13	148	9	53	61	9	53	61	9	53	61	
		6	74	62	9	33	38	8	123	122	1	70	50	10	71	70	10	71	70	10	71	70	
		0	56	29	10	33	38	10	81	81	2	67	70	8	6	183	195	8	6	183	195		
		1	131	119	11	100	114	11	44	48	3	46	37	1	147	163	1	147	163	1	147	163	
		2	107	78	2	40	24	7	2	47	6	133	126	2	80	86	2	80	86	2	80	86	
		3	74	70	3	59	58	1	231	263	6	79	83	3	150	146	3	150	146	3	150	146	
		4	46	38	4	276	298	2	82	71	7	14	0	4	141	142	4	141	142	4	141	142	
		5	112	99	5	36	30	3	204	208	2	68	58	5	150	145	5	150	145	5	150	145	
		6	59	56	6	46	45	4	57	59	3	139	145	6	52	44	6	52	44	6	52	44	
		0	80	73	7	33	30	5	149	164	4	57	47	8	100	82	8	100	82	8	100	82	
		1	80	64	8	118	128	6	71	72	5	81	95	1	41	34	1	41	34	1	41	34	
		2	95	84	9	43	42	8	61	67	7	15	146	2	162	151	2	162	151	2	162	151	
		3	104	99	10	70	77	9	80	79	1	71	45	3	102	108	3	102	108	3	102	108	
		4	50	43	6	6	42	10	34	27	2	76	80	4	87	89	4	87	89	4	87	89	
		5	128	143	2	58	53	11	68	72	3	49	51	5	74	64	5	74	64	5	74	64	
		1	119	102	3	239	250	0	249	254	4	128	129	6	134	118	6	134	118	6	134	118	
		3	88	88	5	59	50	1	66	63	6	72	80	8	160	156	8	160	156	8	160	156	
		4	129	126	6	86	61	2	180	168	0	65	66	1	54	54	1	54	54	1	54	54	
		5	71	75	1	69	74	3	96	101	1	116	91	2	100	99	2	100	99	2	100	99	
		6	77	68	3	37	36	4	189	187	3	83	99	3	47	48	3	47	48	3	47	48	
		0	89	81	4	227	239	5	60	47	4	46	41	4	110	112	4	110	112	4	110	112	
		1	129	104	5	86	74	6	179	178	6	34	43	5	83	71	5	83	71	5	83	71	
		2	47	39	6	50	46	7	57	62	7	17	0	6	60	54	6	60	54	6	60	54	
		3	127	142	1	40	33	8	126	130	1	43	28	7	88	76	7	88	76	7	88	76	
		4	104	88	2	44	36	10	109	111	3	37	43	8	58	45	8	58	45	8	58	45	
		5	66	60	3	123	118	0	35	26	4	377	44	4	76	72	4	76	72	4	76	72	
		6	66	57	4	63	56	1	194	209	2	41	61	5	83	85	5	83	85	5	83	85	
		0	148	182	5	51	53	2	65	78	3	54	55	6	42	37	6	42	37	6	42	37	
		1	101	80	6	60	42	3	195	198	7	19	0	8	90	84	8	90	84	8	90	84	
		2	50	34	4	81	78	4	66	48	1	55	32	1	110	105	1	110	105	1	110	105	
		3	91	93	3	79	63	5	141	136	0	108	116	2	114	118	2	114	118	2	114	118	
		4	130	128	5	43	46	6	83	81	1	168	172	3	112	98	3	112	98	3	112	98	
		5	43	46	4	64	60	7	126	121	2	204	211	4	66	60	4	66	60	4	66	60	
		6	67	61	5	74	65	8	35	31	3	223	244	5	63	68	5	63	68	5	63	68	
		0	94	87	6	95	87	9	59	53	4	81	85	6	81	81	6	81	81	6	81	81	
		1	96	79	0	78	57	10	59	63	5	124	120	7	140	151	7	140	151	7	140	151	
		3	90	95	1	132	128	11	61	66	6	126	120	8	11	67	8	11	67	8	11	67	
		4	59	60	3	40	41	7	119	102	7	136	141	9	91	82	9	91	82	9	91	82	
		5	37	29	4	46	37	1	106	113	9	95	69	4	142	136	4	142	136	4	142	136	
		6	56	51	5	139	172	2	96	79	11	100	92	5	87	103	5	87	103	5	87	103	
		0	84	75	6	52	32	3	103	104	8	1	190	189	6	82	88	6	82	88	6	82	88
		1	88	65	1	75	72	4	111	102	1	99	89	7	80	61	7	80	61	7	80	61	
		3	49	57	2	224	264	5	26	25	2	130	136	8	123	126	8	123	126	8	123	126	
		4	83	64	3	75	83	6	128	122	4	155	168	9	101	123	9	101	123	9	101	123	
		2	49	45	4	54	43	7	66	67	5	149	155	2	123	126	2	123	126	2	123	126	
		3	43	38	6	155	159	8	81	98	6	81	85	3	111	115	6	81	85	3	111	115	
		4	35	28	0	57	52	10	77	85	7	72	79	4	57	43	7	57	43	4	57	43	
		5	54	67	1	177	155	11	32	32	8	114	104	5	83	89	8	114	104	5	83	89	
		0	44	23	3	79	83	1	130	149	9	101	113	6	68	76	9	101	113	6	68	76	
		1	60	52	4	53	38	2	48	50	10	66	63	0	103	123	10	66	63	0	103	123	
		3	36	40	5	153	174	3	115	118	11	46	52	1	59	41	11	46	52	1	59	41	
		0	73	74	6	56	64	5	100	100	8	2	174	167	2	69	55						

TABLE 2 (Continued)

<i>h</i>	<i>k</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	
8	17	4	64	70	4	132	131	9	66	76	5	112	109	12	2	0	180	179						
		0	83	74	5	90	93	10	34	41	6	103	103			1	43	28						
		2	84	96	6	86	93	10	6	68	50	7	80	79			2	65	67					
		3	45	46	9	107	91	1	131	133	8	92	86			4	120	124						
		4	68	69	2	97	86	2	133	129	9	74	72			5	59	54						
		0	55	69	3	88	95	5	58	52	10	73	75			8	56	54						
		1	97	97	4	53	40	4	60	52	4	92	71			9	45	48						
		3	61	65	6	87	95	5	139	157	1	141	138			12	3	0	59	56				
		1	84	77	9	70	70	6	98	103	2	151	153			1	53	30						
		2	69	46	1	100	82	10	7	0	89	62	3	128	120	2	60	40						
		3	59	58	3	87	95	1	90	82	4	88	93			3	46	43						
		4	77	68	4	77	72	2	187	202	5	88	90			4	46	32						
		5	69	72	9	55	50	3	53	55	6	80	70			5	76	77						
		6	119	115	3	45	31	4	47	40	7	50	56			7	83	74						
		8	78	60	4	69	52	5	98	88	8	64	70			8	26	29						
		10	98	104	6	76	76	6	102	104	9	49	51			9	67	81						
		0	94	85	1	108	88	10	8	0	57	35	11	5	0	40	31	12	4	1	40	35		
		1	170	176	2	140	133	1	62	53	1	79	76			2	178	184						
		2	82	74	5	95	104	2	91	64	2	47	42			3	55	50						
		3	127	133	4	69	49	3	47	36	3	67	65			5	40	37						
		4	72	76	6	68	75	5	89	92	4	59	59			6	81	97						
		5	82	78	0	100	81	6	57	52	5	57	54			7	43	51						
		6	74	69	1	137	111	0	140	129	6	79	78			9	23	26						
		7	65	62	2	65	45	1	62	48	8	80	93			12	5	0	96	80				
		8	78	81	3	82	84	2	84	92	9	45	46			1	138	141						
		10	40	39	4	83	79	5	58	59	11	6	0	57	50	3	58	54						
		1	162	165	5	78	103	4	118	99	1	113	122			4	68	77						
		2	77	63	6	55	57	6	39	33	2	42	29			5	131	166						
		3	140	141	9	55	30	0	106	113	3	86	90			8	40	40						
		4	107	110	2	107	133	1	66	37	4	46	41			9	95	126						
		5	113	119	3	81	91	3	106	114	5	73	79			12	6	1	62	50				
		6	97	104	4	51	41	4	100	86	11	7	0	70	43	2	181	211						
		7	79	81	6	65	82	6	69	61	1	95	92			3	74	76						
		8	88	103	1	67	58	10	11	0	162	163	2	48	38	6	133	138						
		9	69	77	3	33	34	1	101	89	3	89	94			12	7	0	90	86				
		10	78	84	4	62	65	3	71	80	4	100	92			1	112	100						
		0	142	138	9	62	46	4	146	149	6	56	47			3	40	23						
		1	200	177	1	80	65	6	42	27	11	8	0	156	148	4	63	64						
		2	200	193	2	37	46	0	107	114	1	141	134			5	119	132						
		3	144	128	3	54	35	1	91	67	2	84	62			6	45	38						
		4	117	127	9	1	35	28	3	111	113	3	88	98			12	8	0	97	72			
		5	115	125	10	0	160	158	4	98	94	4	111	120			1	43	37					
		6	134	132	1	97	92	6	54	50	5	86	91			2	106	115						
		7	70	67	2	87	90	10	13	0	129	153	6	66	69	3	60	68						
		8	93	95	3	160	175	1	97	81	11	9	0	98	94	4	56	58						
		9	46	37	4	122	133	3	56	57	1	128	112			6	75	77						
		10	64	70	6	59	43	4	106	106	3	103	117			12	9	0	39	34				
		0	110	87	7	180	180	10	14	0	47	42	4	95	85	1	34	20						
		1	162	160	8	86	81	3	54	59	5	67	66			3	62	62						
		2	73	65	10	1	206	240	4	54	50	6	88	82			4	34	37					
		3	124	117	1	106	94	10	15	0	72	48	11	10	0	104	124	6	62	59				
		4	115	114	2	79	58	1	87	63	1	127	103			12	10	0	160	158				
		5	120	126	3	137	142	2	51	75	2	52	29			1	66	54						
		6	91	84	4	169	185	3	44	44	3	98	105			3	36	37						
		7	71	80	5	59	44	4	32	32	4	98	100			4	129	133						
		8	99	114	7	81	91	10	16	0	44	30	6	64	55	12	11	1	102	79				
		9	64	64	8	74	80	1	88	61	1	88	61			2	78	63						
		10	58	60	10	57	51	2	70	73	2	70	73			3	128	127						
		0	52	40	10	2	127	104	3	26	25	2	45	39			4	42	30					
		1	144	141	1	44	26	4	43	44	3	64	61			6	50	49						
		2	108	87	3	86	91	10	17	1	76	81	4	46	34	12	12	0	156	167				
		3	113	107	4	92	93	11	0	1	109	97	6	76	77	1	80	67						
		4	51	48	5	60	57	3	121	122	11	12	1	119	89	3	36	35						
		5	87	87	7	114	132	5	56	54	2	67	68			4	96	132						
		6	86	95	8	63	58	6	60	46	5	77	85			6	36	37						
		7	53	52	9	54	57	7	89	60	6	39	58			12	13	1	53	38				
		8	45	45	10	3	134	118	11	1	0	56	37			2	55	55						
		9	36	30	1	49	34	1	78	72	2	64	62			3	72	81						
		10	45	54	2	155	152	2	86	80	3	54	50			4	33	35						
		0	54	33	4	90	88	3	67	69	4	47	39			12	14	1	64	57				
		1	68	62	5	51	39	4	71	70	1	120	96			3	41	37						
		2	46	43	6	80	81	5	70	67	2	95	120			4	54	56						
		3	86	83	7	34	32	6	95	106	3	62	61			12	15	0	63	34				
		4	95	88	8	31	33	7	47	56	4	41	39			1	58	47						
		6	80	66	9	50	46	8	69	70	5	45	35			3	24	20						
		0	110	109	10	36	29	9	50	48	6	56	69			12	16	1	50	47				
		1	120	137	10	4	1	113	107	11	2	0	89	83			13	0	1	44	26			
		2	47	50	2	128	111	1	148	143	4	40	40			2	48	45						
		3	96	99	4	50	42	2	105	99	11	16	1	80	70	3	60	53						
		4	100	98	5	133	141	3																

TABLE 2 (Continued)

<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>
13	3	6	93	96			6	123	128	14	11	0	92	65	15	11	4	29	36
		8	87	99			7	82	88			1	119	102	15	12	1	71	60
		0	40	33			8	54	54			3	60	66	15	13	1	72	39
		1	165	155	14	1	0	111	101			4	52	56	16	0	1	83	63
		2	44	41			1	100	93	14	12	2	80	101			2	125	152
		3	157	149			2	74	82			3	44	41			3	44	39
		5	108	115			3	76	80			4	55	54			5	66	66
		6	45	45			4	83	75	14	13	1	77	65			6	90	86
		7	87	89			5	89	102			3	39	37	16	1	1	84	84
		9	73	72			7	54	49	15	0	1	109	96			2	93	94
13	4	0	84	71	14	2	2	102	103			3	67	69			3	43	30
		1	44	27			3	57	52			5	78	79			5	78	101
		2	132	114			4	36	26			6	49	39			6	64	52
		4	81	83			5	58	63	15	1	1	72	62	16	2	0	46	37
		6	95	100			6	71	66			3	65	63			1	51	37
		8	71	84			7	64	68			4	73	64			2	90	109
13	5	1	120	112	14	3	0	91	94			6	45	35			4	35	30
		3	87	97			2	79	86			0	112	87			5	48	46
		4	48	49			4	68	64			1	112	104			6	61	65
		5	67	61			6	47	50			2	88	95	16	3	0	56	50
		7	44	47			7	51	41			3	78	75			2	46	40
		8	40	52	14	4	0	116	104			4	77	77			3	50	42
13	6	0	56	38			1	42	28			5	68	82			4	40	28
		1	43	46			3	39	38			6	53	59	16	4	0	113	123
		4	71	54			4	80	78	15	3	0	53	25			1	36	30
		6	66	73			5	63	64			1	88	90			2	45	43
13	7	1	117	121			6	48	36			2	50	40			3	45	41
		3	105	103			7	47	57			3	100	106			4	77	87
		5	55	74			8	46	56			4	38	54	16	5	0	71	91
13	8	0	115	131	14	5	0	82	66			5	56	65			1	49	42
		1	47	48			1	94	89			6	61	57			3	73	87
		2	84	94			2	87	97			7	64	71			4	64	75
		3	37	32			3	94	112	15	4	0	77	81			6	23	28
		4	128	127			4	45	43			1	102	102	16	6	1	47	40
		6	83	96			5	55	58			2	64	58			3	63	57
13	9	1	143	133			6	42	45			3	91	83			4	96	105
		2	48	50			7	79	95			4	68	77	16	7	0	73	70
		3	118	131	14	6	0	161	148			5	50	42			1	58	49
		5	75	90			1	47	32			6	51	56			3	74	77
		6	41	34			2	50	63			7	36	41			4	59	67
13	10	0	93	100			4	127	131	15	5	1	37	53			6	25	12
		1	38	26			6	54	56			2	42	44	16	8	0	107	81
		2	58	53			0	98	79			3	42	34			1	45	35
		3	39	36			1	83	76			4	32	29			3	35	30
		4	80	76			2	92	79			6	64	72			4	55	63
		6	63	77			3	88	92			7	32	46	16	9	0	44	25
13	11	1	104	85			4	67	60	15	6	1	90	78			1	44	29
		3	67	69			6	46	42			3	85	88			3	26	9
		4	37	25	14	8	0	94	73			6	32	31			4	35	36
		6	49	44			3	48	43	15	7	1	81	62	16	10	1	44	38
13	12	1	37	27			4	67	68			3	58	54	16	11	0	26	6
		3	48	43			6	27	27			4	52	45			1	86	67
		4	49	39	14	9	0	76	66			6	67	64			2	50	66
13	13	1	125	92			1	51	45	15	8	1	91	74	17	0	1	55	38
		3	55	66			2	48	65			2	83	102			4	58	52
		4	39	36			3	37	35			3	74	74			6	73	49
13	14	1	28	24			4	65	71			4	66	58	17	1	0	62	43
		2	73	88			6	24	34			6	71	74			1	88	82
13	15	3	27	32	14	10	0	63	54			1	105	92			2	31	30
		1	87	83			1	34	13	15	9	4	55	57			3	79	72
14	0	0	68	51			2	78	97			1	88	62			4	35	31
		2	142	149			3	34	37			2	97	78			5	47	58
		3	63	52			4	40	32			3	56	58			6	26	28
		4	56	58			5	48	47			4	32	31	17	2	0	56	44
		5	47	55			6	68	69										

TABLE 3

Bond lengths and their standard deviations (in parentheses) in Å

Pt(1)-O(1)	2.22(0.03)	Pt(2)-O(2)	2.22(0.04)
Pt(1)-O(2)	2.29(0.04)	Pt(2)-O(1)	2.23(0.03)
Pt(1)-N(1)	2.19(0.03)	Pt(2)-N(2)	2.08(0.04)
Pt(1)-C(12)	2.07(0.06)	Pt(2)-C(14)	2.17(0.09)
Pt(2)-C(10)	2.03(0.03)	Pt(2)-C(13)	2.01(0.06)
Pt(1)-C(11)	2.00(0.04)	Pt(2)-C(15)	2.12(0.04)
O(1)-C(1)	1.22(0.05)	O(2)-C(17)	1.27(0.06)
C(1)-C(2)	1.51(0.07)	C(17)-C(18)	1.36(0.06)
C(2)-C(3)	1.45(0.07)	C(18)-C(19)	1.43(0.06)
C(3)-C(4)	1.36(0.07)	C(19)-C(20)	1.30(0.06)
C(4)-C(5)	1.48(0.08)	C(20)-C(21)	1.44(0.06)
C(5)-C(9)	1.43(0.07)	C(21)-C(16)	1.40(0.05)
C(9)-C(1)	1.50(0.06)	C(16)-C(17)	1.52(0.05)
C(5)-C(6)	1.43(0.08)	C(21)-C(22)	1.34(0.07)
C(6)-C(7)	1.25(0.07)	C(22)-C(23)	1.41(0.07)
C(7)-C(8)	1.59(0.07)	C(23)-C(24)	1.55(0.06)
C(8)-N(1)	1.25(0.06)	C(24)-N(2)	1.37(0.06)
C(9)-N(1)	1.33(0.05)	C(16)-N(2)	1.41(0.05)

TABLE 4

Bond angles and their standard deviations

At platinum (all $\pm 1^\circ$)			
O(1)-Pt(1)-O(2)	80°	O(2)-Pt(2)-O(1)	81°
O(1)-Pt(1)-N(1)	76	O(2)-Pt(2)-N(2)	75
O(2)-Pt(1)-N(1)	90	O(1)-Pt(2)-N(2)	90
(all $\pm 2^\circ$)			
C(10)-Pt(1)-C(12)	85	C(14)-Pt(2)-C(13)	94
C(11)-Pt(1)-C(12)	85	C(14)-Pt(2)-C(15)	84
C(10)-Pt(1)-C(11)	90	C(13)-Pt(2)-C(15)	89
O(1)-Pt(1)-C(10)	96	O(2)-Pt(2)-C(13)	93
O(1)-Pt(1)-C(11)	98	O(2)-Pt(2)-C(15)	101
O(2)-Pt(1)-C(12)	99	O(1)-Pt(2)-C(14)	92
O(2)-Pt(1)-C(11)	91	O(1)-Pt(2)-C(15)	91
N(1)-Pt(1)-C(12)	102	N(2)-Pt(2)-C(14)	102
N(1)-Pt(1)-C(10)	89	N(2)-Pt(2)-C(13)	89
At oxygen			
Pt(1)-O(1)-Pt(2)	$99 \pm 1^\circ$	Pt(2)-O(2)-Pt(1)	$97 \pm 1.5^\circ$
Pt(1)-O(1)-C(1)	113 ± 3	Pt(2)-O(2)-C(17)	113 ± 3
Pt(2)-O(1)-C(1)	111 ± 3	Pt(1)-O(2)-C(17)	119 ± 3
In the 8-quinolinolate groups (all $\pm 4^\circ$)			
O(1)-C(1)-C(2)	128°	O(2)-C(17)-C(18)	122°
O(1)-C(1)-C(9)	120	O(2)-C(17)-C(16)	119
C(9)-C(1)-C(2)	113	C(16)-C(17)-C(18)	118
C(1)-C(2)-C(3)	121	C(17)-C(18)-C(19)	120
C(2)-C(3)-C(4)	124	C(18)-C(19)-C(20)	121
C(3)-C(4)-C(5)	119	C(19)-C(20)-C(21)	124
C(4)-C(5)-C(6)	119	C(20)-C(21)-C(22)	130
C(4)-C(5)-C(9)	119	C(20)-C(21)-C(16)	117
C(6)-C(5)-C(9)	122	C(22)-C(21)-C(16)	114
C(5)-C(6)-C(7)	114	C(21)-C(22)-C(23)	128
C(6)-C(7)-C(8)	124	C(22)-C(23)-C(24)	109
C(7)-C(8)-N(1)	115	C(23)-C(24)-N(2)	127
C(8)-N(1)-C(9)	120	C(24)-N(2)-C(16)	109
Pt(1)-N(1)-C(8)	124	Pt(2)-N(2)-C(24)	129
Pt(1)-N(1)-C(9)	111	Pt(2)-N(2)-C(16)	119
N(1)-C(9)-C(5)	119	N(2)-C(16)-C(21)	131
N(1)-C(9)-C(1)	117	N(2)-C(16)-C(17)	109
C(1)-C(9)-C(5)	124	C(17)-C(16)-C(21)	120

TABLE 5

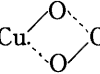
Deviations of the atoms from planes (Å)

Atoms included in calculating the plane:							
I		II		III		IV	
O(1)	-0.08	O(2)	+0.10	C(10)	-0.20	C(13)	-0.05
N(1)	+0.10	N(2)	-0.01	C(12)	+0.02	C(14)	+0.05
C(1)	-0.01	C(17)	-0.07	O(1)	+0.01	O(1)	-0.04
C(2)	0.00	C(18)	-0.02	O(2)	-0.01	O(2)	+0.04
C(3)	-0.03	C(19)	-0.05				
C(4)	+0.03	C(20)	+0.05				
C(5)	+0.02	C(21)	0.00				
C(6)	-0.01	C(22)	-0.01				
C(7)	-0.12	C(23)	+0.12				
C(8)	0.00	C(24)	-0.13				
C(9)	+0.10	C(16)	+0.01				
Atoms not included in calculating the plane:							
I		II		III		IV	
Pt(1)	+0.41	Pt(2)	-0.40	Pt(1)	-0.03	Pt(2)	-0.04
Pt(2)	-2.19	Pt(1)	+2.22				
Equations of the planes							
I	-0.749X	-0.626Y	-0.218Z	+2.460	=	0	
II	-0.898X	-0.178Y	-0.403Z	+3.169	=	0	
III	-0.392X	+0.827Y	-0.403Z	-1.992	=	0	
IV	-0.275X	+0.929Y	-0.247Z	-2.692	=	0	

equivalent but crystallographically independent bonds; the average difference both in the light-atom-light-atom and the platinum-light-atom bond lengths is, as shown in Table 3, 0.07 Å, which is the same as the average standard deviation for the bonds involving only light atoms, but twice that for the platinum-light-atom bonds. This suggests, quite reasonably, that it is for the atoms directly attached to platinum that the parameter interaction is largest and for which the standard deviations have been underestimated by the block diagonal refinement.

The weighted mean Pt-C(methyl) bond length is 2.06 Å, in agreement with values found in other determinations; ^{1,2} the Pt-N bond length, 2.13 Å, is also the same as that found in Me₃ acac bipy Pt.² In several compounds (*e.g.*, dimolybdenum decachloride⁸) the bond length metal-X(bridging) has been found to be greater than metal-X(non-bridging). The Pt-O bond lengths in this complex appear to follow this trend, being 0.07 Å longer than those in the β-diketone complexes,¹ but the difference is not statistically significant. The ready fission of the Pt-O' bond shown by the experiments of Kite and Truter⁴ is not reflected in any significant difference between the Pt-O and the Pt-O' bond lengths.

Other 8-hydroxyquinoline complexes have been studied in detail by Palenik.⁹ He found the bond lengths in the chelated 8-quinolinolato-ligand to be the same in the β-form of anhydrous bis-(8-quinolinolato)copper(II) and in bis-(8-quinolinolato)diaquozinc, although in the former compound one 8-quinolinolate forms an unsymmetrical

Cu  Cu bridge while in the latter compound there is no bridging. Within our larger

experimental error, the bond lengths in the 8-quinolinolate ligand, with a symmetrical bridge, are the same as those in the copper and zinc compounds. Palenik found that the oxygen taking part in bridging was displaced from the plane of the quinoline ring towards the second copper atom. From Table 5 it can be seen that there is apparently a similar effect but the displacements of the oxygen atoms are only three times their minimum standard deviations and the displacements of the other atoms do not follow a logical pattern in the halves of the dimer. The platinum atoms are undoubtedly not coplanar with the 8-quinolinolate ligands.

The strain in the formation of a four-membered ring is shared by the platinum and oxygen atoms; all have bond angles 10° less than the ideal values (90° at Pt and 109.5° at O). The angle at the platinum atom in the chelate ring, 75°, is also significantly less than 90°; comparison with other 8-quinolinolates⁹ suggests that this results from the larger covalent radius of platinum, the N · · · O distance being similar in all the compounds. The angles involving the methyl groups are all 90°. As in the other trimethylplatinum complexes, the atom forms three short Pt-C bonds mutually at right angles, and three relatively longer bonds the angles among which are 90° or less.

EXPERIMENTAL

The colourless crystals were prepared by Kite and Truter.⁴ Microscopic investigation, particularly the appearance of etch pits indicated that they belonged to a polar crystal class. Preliminary examination by X-rays at room temperature gave the space group and approximate unit-cell dimensions from which the number of formula units was deduced.

For intensity measurements the crystal was cooled in a stream of gaseous nitrogen with an apparatus adapted from that described by Robertson.¹⁰ No phase transformation took place on cooling. The temperature was measured from time to time and found to be 120° ± 5°K. Accurate values of the cell dimensions were obtained from Al-calibrated Weissenberg photographs. Equant crystals of side 0.17–0.19 mm. were used. Equi-inclination Weissenberg

⁸ D. E. Sands and A. Zalkin, *Acta Cryst.*, 1959, **12**, 723.

⁹ G. J. Palenik, *Acta Cryst.*, 1964, **17**, 687, 696.

¹⁰ J. H. Robertson, *J. Sci. Instr.*, 1960, **37**, 41.

photographs were taken about the three principal axes up to $3kl$, $h5l$, and $hk6$; 2152 of a possible 2624 reflections gave measurable intensities. Intensities were estimated visually with the aid of a calibration strip. Lorentz, polarisation, and absorption (as for a sphere of radius 0.09 mm.) corrections were applied, and then the individual layers were correlated to give a set of relative structure amplitudes; these calculations were carried out on the Leeds University Pegasus Computer with programmes written by J. G. F. Smith. The scale factor was one of the parameters in the refinement.

Structure factors and least-squares refinement calculations were carried out on the Pegasus computer with programmes written by Cruickshank and Pilling.¹¹ The function minimised was $R = \Sigma w(|F_o| - |F_c|)^2$ where the weighting factor, w , was $1/|F_o|$. For platinum we used the scattering factor of Thomas and Umeda,¹² corrected for the real part of the anomalous dispersion;¹³ for the other atoms we used the Berghuis *et al.*¹⁴ scattering factors, all for neutral atoms. Three dimensional Fourier difference syntheses were carried out on the computer.¹¹

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DEPARTMENT OF INORGANIC AND STRUCTURAL CHEMISTRY,
THE UNIVERSITY, LEEDS 2.

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¹¹ D. W. J. Cruickshank, D. E. Pilling, and (in part) A. Bujosa, F. M. Lovell, and M. R. Truter, in "Computing Methods and the Phase Problem in X-ray Crystal Analysis," Pergamon Press, London, 1961, p. 32.

¹² L. H. Thomas and K. Umeda, *J. Chem. Phys.*, 1957, **26**, 293.

¹³ C. H. Dauben and D. H. Templeton, *Acta Cryst.*, 1955, **8**, 841.

¹⁴ J. Berghuis, I. M. Haanappel, M. Potters, B. O. Loopstra, C. H. MacGillavry, and A. L. Veenendaal, *Acta Cryst.*, 1955, **8**, 478.
